Table 2. Selected geometric parameters ( $\AA^{\circ},^{\circ}$ )

| N1-C2 | $1.343(4)$ | $\mathrm{C} 5-\mathrm{C} 6$ | $1.493(6)$ |
| :--- | ---: | :--- | ---: |
| $\mathrm{C} 2-\mathrm{C} 5$ | $1.502(5)$ | $\mathrm{N} 2-\mathrm{C} 11$ | $1.411(5)$ |
| $\mathrm{C} 5-\mathrm{N} 2$ | $1.274(5)$ |  |  |
| $\mathrm{N} 2-\mathrm{C} 5-\mathrm{C} 6$ | $126.5(3)$ | $\mathrm{C} 6-\mathrm{C} 5-\mathrm{C} 2$ | $117.9(3)$ |
| $\mathrm{N} 2-\mathrm{C} 5-\mathrm{C} 2$ | $115.6(3)$ |  |  |
| $\mathrm{N} 1-\mathrm{C} 2-\mathrm{C} 5-\mathrm{N} 2$ | $-178.7(4)$ | $\mathrm{C} 13-\mathrm{C} 14-\mathrm{Ol}-\mathrm{C} 17$ | $0.3(6)$ |

Molecule (I) crystallized in the orthorhombic system; space groups $C m c 2_{1}, C m c m$ or $C 2 \mathrm{~cm}$ allowed by the systematic absences. With $Z=4$, space group Cmcm requires that the molecule have either mm or $2 / \mathrm{m}$ symmetry; this is not possible and this space group option was rejected. Of the two remaining space groups (both of which require that the molecule have crystallographic $m$ symmetry), only the $C m c 2_{1}$ choice afforded a solution. The molecule is achiral (has mirror symmetry) and with only $\mathrm{C}, \mathrm{H}, \mathrm{N}$ and O atoms (which have very small anomalous dispersion components) we deemed it very unlikely that a correct determination of the direction of the polar axis in the $C m c 2_{1}$ space group would be possible. As a consequence of this decision, we measured one unique octant of data to a $\theta_{\text {max }}$ of $27.5^{\circ}$ with Mo $K \alpha$ radiation and obtained a reflection/parameter ratio of 9.18. As anticipated, the refinement did not allow a determination of the chiral axis direction.

Data collection: CAD-4-PC Software (Enraf-Nonius, 1992). Cell refinement: SET4 and CELDIM in CAD-4-PC Software. Data reduction: DATRD2 in NRCVAX96 (Gabe, Le Page, Charland, Lee \& White, 1989). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985). Program(s) used to refine structure: NRCVAX96 and SHELXL93 (Sheldrick, 1993). Molecular graphics: NRCVAX96, ORTEPII (Johnson, 1976), PLATON (Spek, 1996a) and PLUTON (Spek 1996b). Software used to prepare material for publication: NRCVAX96, SHELXL93 and WordPerfect macro PREPCIF (Ferguson, 1996).

ECA and GF thank NSERC (Canada) for research grants.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: AB1460). Services for accessing these data are described at the back of the journal.

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Acta Cryst. (1997). C53, 890-892

## $N$-(6-Amino-3,4-dihydro-3-methyl-5-nitroso-4-oxopyrimidin-2-yl)glycine-Water (1/2)

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(Received 17 February 1997; accepted 10 March 1997)

## Abstract

The title compound, $\mathrm{C}_{7} \mathrm{H}_{9} \mathrm{~N}_{5} \mathrm{O}_{4} \cdot 2 \mathrm{H}_{2} \mathrm{O}$, makes use of the two waters of crystallization to form a hydrogenbonded network which utilizes all available $\mathrm{N}-\mathrm{H}$ and $\mathrm{O}-\mathrm{H}$ groups as hydrogen-bond donors. The shortest intermolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ bond is 2.491 (3) $\AA$.

## Comment

The title compound, (I), is of potential biological interest since it is an adduct of the amino acid glycine and a substituted uracil nucleobase. In addition, this type of compound can coordinate with transition metal ions such as $\mathrm{Cd}^{\text {II }}$ and $\mathrm{Hg}^{\text {II }}$, whose complexes we are currently investigating.

(I)

A view of the asymmetric unit is shown in Fig. 1 and details of the geometry are given in Table 1. The dimensions of the $\mathrm{N} 2-\mathrm{C} 2-\mathrm{N} 1-\mathrm{C} 6-\mathrm{N} 6$ frag-
ment indicate a delocalization of charge with $\mathrm{C}-\mathrm{N}$ distances in the range 1.311 (3)-1.332 (3) $\AA$. The nitroso bond lengths (values in brackets) differ from those found in 4-amino-5-nitroso-2-methoxy-1-methyl6 -oxo-1,6-dihydropyrimidine (Low et al., 1994), C5N5 1.316(3) [1.352(2) A] and N5-O5 1.296(2) [1.266(2) A], whereas the bond angle O5-N5-C5 $\left.117.6(2)\left[117.8(2)^{\circ}\right)\right]$ is almost identical. The nitroso O5 atom takes part as an acceptor in an intramolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bond with the adjacent 6 -amino group to form a planar six-membered ring; this conformation is maintained in six-deuterated DMSO solution, as shown by ${ }^{1} \mathrm{H}$ NMR, where the corresponding signals to both H atoms attached at N6 are well differentiated (chemical shift for H61 and H62 are 10.93 and 8.31 p.p.m., respectively). The mean plane of glycyl atoms $\mathrm{C} 21, \mathrm{C} 22, \mathrm{O} 21$ and O 22 is inclined at $86.0(1)^{\circ}$ to the mean plane of the uracil ring and its substituents ( N 2 , $\mathrm{C} 3, \mathrm{O} 4, \mathrm{~N} 5, \mathrm{O} 5, \mathrm{~N} 6$ ). The uracil moiety and atoms N 2 , C3, O4, N5, O5 and N6 lie in a plane, with atoms N2 and N6 showing the largest deviations from the plane at distances of -0.078 (2) and -0.074 (2) $\AA$, respectively. The largest deviations from the mean plane of the ring atoms themselves are $\mathrm{C} 2-0.017$ (2) and $\mathrm{C} 60.022(2) \AA$.


Fig. 1. The molecular structure of the title compound with the numbering scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level (ORTEPII; Johnson, 1976).

Molecules of the title compound are linked by strong $[2.491$ (3) $\AA$ ] $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds [ $\mathrm{O} 22-$ $\mathrm{H} 22 \cdots \mathrm{O}^{\vee} 2.491$ (3) $\AA$; see Table 3] to form zigzag
chains which extend in the b direction. These zigzag chains are linked to adjacent chains to form puckered sheets which run parallel to the $c$ axis via $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$; $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ and $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds involving three of the four H atoms of the two water molecules. The puckered sheets are linked to form a network via $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds involving the fourth water H atom $\left[\mathrm{Ol}-\mathrm{H} 1 A \cdots \mathrm{O}^{\mathrm{i}}, \mathrm{O} \cdots \mathrm{O} 2.779\right.$ (3) $\AA$ A]. Examination of the structure with PLATON (Spek, 1995a) showed that there are no solvent-accessible voids in the crystal lattice.

## Experimental

The title compound was prepared by adding 6 -amino- 3,4 -di-hydro-3-methyl-2-methoxy-5-nitroso-4-oxopyrimidine ( 2.022 g , 10.0 mmol ) to a solution of glycine $(0.751 \mathrm{~g}, 10.0 \mathrm{mmol})$ in 32 ml of 0.31 M aqueous $\mathrm{KOH}(9.92 \mathrm{mmol})$. The mixture was stirred at 343 K for 50 min . The wine-red solution was cooled and the pH was adjusted to 3.0 by dropwise addition of glacial acetic acid to neutralize and induce crystallization. After 5 h at room temperature, the orange crystalline solid was collected by filtration and washed successively with water, ethanol and diethyl ether to yield the dihydrate title compound $(2.125 \mathrm{~g}$, $7.27 \mathrm{mmol} ; 81 \%$ yield).

## Crystal data

$\mathrm{C}_{7} \mathrm{H}_{9} \mathrm{~N}_{5} \mathrm{O}_{4} .2 \mathrm{H}_{2} \mathrm{O}$
$M_{r}=263.22$
Orthorhombic
Pbca
$a=9.2739$ (13) $\AA$
$b=16.682(2) \AA$
$c=14.690$ (2) $\AA$
$V=2272.6(5) \AA^{3}$
$Z=8$
$D_{x}=1.539 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ not measured

## Data collection

Enraf-Nonius CAD-4
diffractometer
$\theta / 2 \theta$ scans
Absorption correction: none
2485 measured reflections
2485 independent reflections
1207 reflections with $I>2 \sigma(I)$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.048$
$w R\left(F^{2}\right)=0.121$
$S=0.889$
2485 reflections
177 parameters
H atoms riding: see below
$(\Delta / \sigma)_{\max }=-0.001$
$\Delta \rho_{\text {max }}=0.246 \mathrm{e}^{-3}$
$\Delta \rho_{\text {min }}=-0.272 \mathrm{e}^{-3}$
Extinction correction: none
Scattering factors from International Tables for Crystallography (Vol. C)

Mo $K \alpha$ radiation
$\lambda=0.7107 \AA$
Cell parameters from 25 reflections
$\theta=8.55-14.55^{\circ}$
$\mu=0.134 \mathrm{~mm}^{-1}$
$T=294$ (2) K
Plate
$0.41 \times 0.41 \times 0.03 \mathrm{~mm}$
Orange

$$
\begin{aligned}
& \theta_{\max }=27.0^{\circ} \\
& h=0 \rightarrow 11 \\
& k=0 \rightarrow 21 \\
& l=0 \rightarrow 18 \\
& 3 \text { standard reflections } \\
& \quad \text { frequency: } 120 \mathrm{~min} \\
& \text { intensity variation: } 0.4 \%
\end{aligned}
$$

$w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0614 P)^{2}\right]$
where $P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3$

Table 1. Selected geometric parameters $\left(\AA^{\circ},^{\circ}\right)$

| $\mathrm{N} 1-\mathrm{C} 2$ | $1.323(3)$ | $\mathrm{N} 3-\mathrm{C} 3$ | $1.468(3)$ |
| :--- | ---: | :--- | ---: |
| $\mathrm{N} 1-\mathrm{C} 6$ | $1.332(3)$ | $\mathrm{N} 3-\mathrm{C} 4$ | $1.381(3)$ |
| $\mathrm{C} 2-\mathrm{N} 2$ | $1.319(3)$ | $\mathrm{C} 4-\mathrm{O} 4$ | $1.210(3)$ |
| $\mathrm{C} 2-\mathrm{N} 3$ | $1.376(3)$ | $\mathrm{C} 4-\mathrm{C} 5$ | $1.469(3)$ |
| $\mathrm{N} 2-\mathrm{C} 21$ | $1.447(3)$ | $\mathrm{C} 5-\mathrm{N} 5$ | $1.316(3)$ |
| $\mathrm{C} 21-\mathrm{C} 22$ | $1.497(3)$ | $\mathrm{C} 5-\mathrm{C} 6$ | $1.444(3)$ |
| $\mathrm{C} 22-\mathrm{O} 21$ | $1.207(3)$ | $\mathrm{N} 5-\mathrm{O} 5$ | $1.296(2)$ |
| $\mathrm{C} 22-\mathrm{O} 22$ | $1.292(3)$ | $\mathrm{C} 6-\mathrm{N} 6$ | $1.311(3)$ |
| $\mathrm{C} 2-\mathrm{N} 2-\mathrm{C} 21-\mathrm{C} 22$ | $-81.3(3)$ | $\mathrm{C} 6-\mathrm{C} 5-\mathrm{N} 5-\mathrm{O} 5$ | $0.0(4)$ |

Table 2. Hydrogen-bonding geometry $\left(\AA^{\circ},^{\circ}\right)$

| $D — \mathrm{H} \cdots \mathrm{A}$ | D-H | H $\cdots$ A | D. ${ }^{\text {d }}$ | D-H. ${ }^{\text {c }}$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O} 1-\mathrm{H} 1 \mathrm{~A} \cdots \mathrm{O} 2$ | 0.82 | 1.97 | 2.779 (3) | 169 |
| $\mathrm{O} 1-\mathrm{H} 1 \mathrm{~B} \cdot . \mathrm{O} 21^{\text {i }}$ | 0.82 | 1.99 | 2.784 (3) | 163 |
| $\mathrm{N} 2-\mathrm{H} 2 \cdots \mathrm{O} 1^{\text {ii }}$ | 0.86 | 2.09 | 2.862 (3) | 149 |
| $\mathrm{O} 2-\mathrm{H} 2 \mathrm{~A} \cdots \mathrm{O} 4^{\text {iii }}$ | 0.82 | 2.39 | 2.901 (3) | 121 |
| O2-H2A $\cdots \mathrm{N} 5^{\text {iii }}$ | 0.82 | 2.46 | 3.228 (3) | 158 |
| $\mathrm{O} 2-\mathrm{H} 2 \mathrm{~B} \cdot \cdots \mathrm{O} 1^{\text {iv }}$ | 0.82 | 2.10 | 2.908 (3) | 170 |
| N6-H6A . . O 2 | 0.86 | 2.07 | 2.914 (3) | 169 |
| N6-H6B..O5 | 0.86 | 2.02 | 2.645 (3) | 129 |
| N6-H6B $\cdots \mathrm{O} 4^{\text {iii }}$ | 0.86 | 2.50 | 2.900 (3) | 109 |
| O22-H22. ${ }^{\text {O }}{ }^{\text {v }}$ | 1.10 | 1.44 | 2.491 (3) | 155 |
| O22-H22 . ${ }^{\text {N }}{ }^{\text {v }}$ | 1.10 | 2.08 | 2.902 (3) | 129 |

Symmetry codes: (i) $x-\frac{1}{2}, y, \frac{1}{2}-z$; (ii) $\frac{1}{2}+x,-\frac{1}{2}-y, 1-z$; (iii) $\frac{3}{2}-x,-y, z-\frac{1}{2}$; (iv) $\frac{1}{2}+x, y, \frac{1}{2}-z$; (v) $\frac{3}{2}-x, y-\frac{1}{2}, z$.

H atoms were placed in calculated positions, except for H 22 , which was located in a $\Delta \rho$ map, $1.10 \AA$ from O 22 , and held in that position. They were treated as riding atoms in the refinement ( $\mathrm{C}-\mathrm{H} 0.93-0.98, \mathrm{~N}-\mathrm{H} \mathrm{0.86}$, and $\mathrm{O}-\mathrm{H} 0.82$ and $1.10 \AA$ ).
Data collection: CAD-4-PC (Enraf-Nonius, 1992). Cell refinement: SET4 and CELDIM in CAD-4-PC. Data reduction: DATRD 2 in NRCVAX 94 (Gabe, Le Page, Charland, Lee \& White, 1989). Program(s) used to solve structure: SOLVER in NRCVAX94. Program(s) used to refine structure: NRCVAX94 and SHELXL 93 (Sheldrick, 1993). Molecular graphics: NRCVAX94, PLATON and PLUTON (Spek, 1995b). Software used to prepare material for publication: NRCVAX94, SHELXL93 and WordPerfect macro PREPCIF.

GF thanks NSERC (Canada) for Research Grants.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: AB1473). Services for accessing these data are described at the back of the journal.

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Acta Cryst. (1997). C53, 892-895
trans- and cis-S—C-C-S Conformations in 5 -( $2,2^{\prime}$-Dithien- 5 -yl)-2 $\mathbf{2}^{\prime}$-deoxyuridine

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#### Abstract

The title compound, $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}_{2}$, adopts a transoidcisoid disordered crystal structure of the terminal thienyl group at room temperature, with occupancies of 0.55 and 0.45 . The structure was solved by direct methods and refined by least-squares calculations to a final $R$ of 0.048 . Molecular cohesion is stabilized by hydrogen bonding involving the two hydroxyl groups of the sugar moiety and the uridine ring.


## Comment

As part of our attempts to identify antiherpes agents by rational drug-design approaches, the structures of a series of deoxyuridine analogues have been characterized (Creuven et al., 1996; Olivier et al., 1994, and references therein). In this contribution, we present the crystal structure of 5-( $2,2^{\prime}$-dithien-5-yl)-2'-deoxyuridine, (I), a compound that shows moderate activity against herpes simplex virus type 1 (Luyten et al., 1995).

(I)

The molecular structure with the atom-numbering scheme is given in Fig. 1. The observed geometry corresponds to standard values (Allen et al., 1987) and is similar to that of previously reported structures for analogues of 5 -substituted $2^{\prime}$-deoxyuridines (Olivier et al., 1994; Creuven et al., 1996).

